www.ofi.at





Influence of the PV-Module design and composition on the formed UV-Fluorescence patterns

Gabriele C. Eder and Yuliya Voronko,

OFI, Austrian Research Institute for Chemistry and Technology, Vienna Karl Knöbl

University of Applied Science-Technicum Vienna



Reliability and durability of photovoltaic modules are a **key factor** for the development of emerging PV markets worldwide

> Reliability is directly dependent on the **chemical and physical stability** of the polymeric encapsulation materials

> > One method capable of **detecting ageing effects of the polymeric encapsulant** directly in the field is

UV Fluorescence imaging







- Excitation: UV-light source
- **Detection**:

(i) eye or photographic camera

 \rightarrow image creating method

(ii) a probe connected to an UV/VIS spectrometer via an optical fibre

→ **spectroscopy** visualising the downshift of the reflected light





Equipment – UV-F Imaging

UV-Fluorescence Imaging

- non-invasive
- easy to handle (portable power supply)
- fast (exposure time ~ 30sec)
- ✓ UV-light source : three power- tuneable LED-arrays with an emission maximum at 365nm (50W, tunable) & low pass filter to cut off visible light



- Detection of UV-F: digital photographic camera (Olympus OM D) equipped with high pass filter to cut-off the UV-irradiation)
- Power supply : a modified DC/DC converter, sourced by a twelve cell Lithium-polymeraccumulator with a capacity of 5000 mAh



Equipment UV-F Spectroscopy

UV-Fluorescence Spectroscopy

- non-destructive
- easy to handle
- outdoor & indoor measurements possible



✓ UV-light source

LED with an emission maximum at 365nm

✓ Detection of UV-F

UV/VIS spectrometer Ocean Optics MAYA 2000 Pro (200-1100nm) with optical fibre ocean optics QR600-7-SR 125 BX

✓ Power supply: two cell Lithium-polymer-accumulator with a capacity of 1600 mAh for LED and notebook for spectrometer





Fluorescence patterns / UV-F images





Influencing stress factors

In the field, PV-modules are exposed to various climatic stress factors like irradiance, temperature, humidity, ...







Formation of Fluorescence

Non-radiative S₁ transition Absorption Fluorescence Energy **S**₀ 2 Ground State

https://en.wikipedia.org/wiki/Jablonski_diagram

Fluorescence: emission of light by a material that has absorbed light or other electromagnetic radiation λ emitted light > λ adsorbed light

Fluorophore : fluorescent chemical compound that can re-emit light upon light excitation; mostly contains delocalized electrons (e.g. combined aromatic groups, plane or cyclic molecules with several π bonds or polymers in excited-state structure) Typically: aeging/degradation products of additives and/or polymers or precursors thereof: *"activated molecules"*

Formation (different types) in the absence of $O_2^{:}$

- Irradiation and elevated T (outdoors)
- high humidity and T (e.g. DH)
- high T (e.g. hot cells)



$natural \leftrightarrow artificial ageing$



- DH-aging: UV-F above the backsheet and at outer regions of the cells
- CL-irradiation: homogeneous fluorescence over cells, extinctions at the rims
- Natural ageing: UV-F pattern comparable to the combined aging CL+I



Effect of stress factors





- peak ~480 nm fluorescence generated upon environmental stress (irradiation, temperature, humidity)
- ✓ second peak ~550nm additional thermal stress in the encapsulate (hot cells in thermographic image)

www.ofi.at



Fluoresence = Polymer Degradation? NO

Comparative analysis of EVA in parts with bright and dark fluorescence within a naturally aged PV-module:

Infrared spectroscopic analysis of EVA

No spectral differences caused by chemical degradation reactions detectable; only small shifts in wavenumber of some bands (e.g. changes in molecular environment)

Thermal analysis (DSC) of EVA

No indicators for chemical polymer degradation detectable but in the low temperature region of the first melting curve (physical effects e.g. crystallinity, orientation..)

TD / GC-MS analysis of EVA

No differences in additive concentration or degradation products detectable





Extinction of Fuorescence

Quenching: decrease of fluorescence intensity of a given substance caused by e.g. energy transfer, complex-formation or collisions. Typical quencher: oxygen (reversible)

(Photo)-Bleaching: decrease of fluorescence intensity due to (photo)-chemical alteration of the fluorophore e.g. by cleaving of covalent bonds with surrounding molecules (permanent)

→ extinction at the rims of the cells increases with module age; extinction above cell-cracks



Fluorescence extinction

Fluorescence is extinct in the presence of oxygen AND irradiation **Photo-bleaching** Polymer is chemically altered = degradation

ORTSCHRITTIN

Temperature accelerates the process





Modell for the extinction of fluorescence

- 1. After outdoor weathering fluorescence effects are observed in the encapsulation of PV-module
- With increasing exposure time, the polymer ages (formation of activated sites; excited-state structure) due to the influence of various stress factors and operation conditions -> fluorescence effects are observed
- 3. The fluorescence of materials can be extinguished by "quenching"-effects which lead to a decrease of fluorescence at the rims of the cells due to non-radiating processes like collision/energy transfer with e.g. permeating oxygen; the polymer is chemically not altered.
- 4. An extinction of strong fluorescence above the cells can be achieved in the presence of oxygen AND irradiation. Elevated temperatures accelerate the process (photo-bleaching). The resulting polymer has modified polymer chain structure which leads to different thermo-mechanical properties.



....apart from stress factors applied

Barrier-properties of the backsheet

- Oxygen, Water Vapour ,...-> quenching and bleaching effects
- Glass / breathable↔ high barrier polymeric composite BS

Type and composition of the encapsulant

- Additives
- Aeging/degradation effects (hydrolysis, photo-oxidation..)

Glued (ECA) ↔ soldered ribbons

- Fluorescing components of ECA \rightarrow permeation
- Material incompatibilities



Influence of barrier properties on fluorescence pattern

Gas-permeable backsheet

- oxygen
- water vapour
- decomposition products e.g.
 EVA: acetic acid



- Oxygen and other molecules permeating into the front encapsulant via a
 permeable backsheet can quench/bleach the fluorescence in the interspaces
 between the cells and above cell cracks → specific fluorescence pattern
- the extent of fluorescence extinction is dependent on OTR of the backsheet and the encapsulant and the exposure time



Permeation Properties of Backsheets



Oxygen Permeation



O₂ is important in fluoresence extinction

www.ofi.at

Effect of type of Backsheet on UV-F pattern

Fluorescence pattern of naturally aged (~5 years, same plant) PV modules with different backsheets



TSCHRITT IN



The OTR through the backsheet into the encapsulation determines the extent of the "extinction of the fluorescence" = **Oxygen-bleaching**;

- \rightarrow broad rims with no fluorescence (extinguished) around the cells;
- \rightarrow cell-cracks become visible



Differences in the UV-F Pattern

Besides the **type of backsheet (barrier-properties)** also **different qualities/types of encapsulant** can cause varying fluorescence effects

- the **polymer-type**
- impurities
- crosslinker and its decomposition products
- additives (and their degradation products)
 can show fluorescence*.

^{*} C. Peike et al, "Towards the origin of photochemical EVA discoloration", Proceedings of the IEEE · June 2013; DOI: 10.1109/PVSC.2013.6744447



UV-F of different native encapsulants

Identical backsheet varying encapsulation



TPO

POE (crosslinked)





Identical backsheet varying encapsulation

EVA / Peroxidic crosslinker, AO, UVA-Absorber

TPO / no crosslinker, no additives in front-layer

POE / chemical crosslinker



In general, the polymeric encapsulate of PV modules does not show distinctive fluorescence effects in the original state. This holds true for the polymer itself, however, special constituents (impurities, additives or fragments of a cross linker) exhibit some weak effects which mostly depend on the exact encapsulant formulation.

With increasing lifetime, the encapsulant material ages due to the interaction with climatic stress factors (mainly irradiation by sunlight and elevated temperature) by forming fluorophores. The fluorescing effect is increasing with exposure time and is typically observable for PV-modules in operation in the field for more than a year.

Example: Visulaisation of compound migration

HAN



Example: visualisation of material incompatibility



TPO Glas/ Glass Modules Solder Sn/Pb



Example: visualisation of water vapour ingress



UV-Fluorescence images of test modules

- (a) in the original state,
- (b) after 1000h
- (c) after 2000h
- (d) after 3000h
- of exposure to DH at 85°C/85% r.H.
- ightarrow visualisation of water ingress / hydrolysis of EVA



Summary: Influence on fluorescence

It was found that the **fluorescence image patterns** obtained for the naturally and/or artificially aged modules depend

- (i) on the composition of the PV-module; in particular if gas-tight (glass, backsheets with Al-barrier layers) or breathable (polymeric backsheets) are used (Oxygen Quenching)
- (ii) If cell-cracks, pinholes in the cell or hot cells are present (Oxygen Q.)
- (iii) on the stress factors applied

The fluorescence spectra (intensity, maxima) depend

- (ii) the type of encapsulant used especially on the additive mixture and peroxidic curing system used
- (iii) on the stress factors (temperature, humidity, irradiation,) applied at the installation site or in the climatic chamber in accelerated ageing tests



Thank you for your attention !

CONTACT: Gabriele Eder phone: +43 1 798 1601 250 mail: gabriele.eder@ofi.at www.ofi.at







This work was founded by the Austrian Klima-& Energiefonds under the projects "INFINITY", FFGnr.: 848771 and AMSEL FFGnr.:848771

